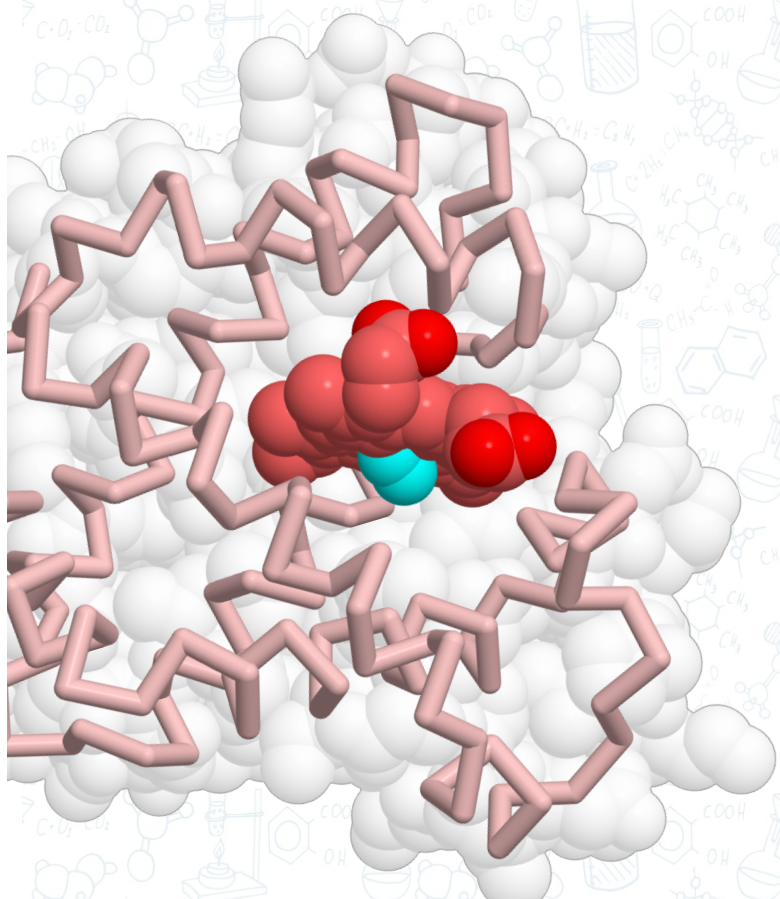


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Half-Sandwich Dioxygen Complexes of Ruthenium: Preparation and Reactivity

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Dioxygen complexes $[\text{Ru}(\eta^5\text{-C}_5\text{Me}_5)(\eta^2\text{-O}_2)\{\text{P}(\text{OR})_3\}\text{L}]\text{BPh}_4$ (**1–3**) were prepared by reacting chloro-compounds $\text{RuCl}(\eta^5\text{-C}_5\text{Me}_5)\{\text{P}(\text{OR})_3\}\text{L}$ with air (1 atm) in the presence of NaBPh_4 .

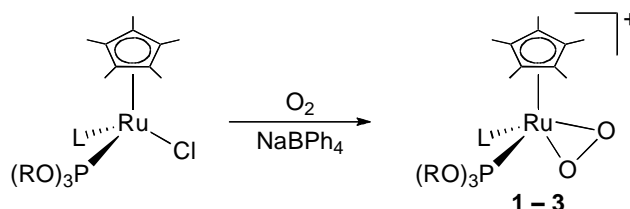


Figure 1: $R = \text{Me}, \text{Et}$; $L = \text{P}(\text{OR})_3, \text{PPh}_3$

Protonation of complexes **1–3** with triflic acid yielded phosphate complexes $[\text{Ru}(\kappa^1\text{-OTf})(\eta^5\text{-C}_5\text{Me}_5)\{\text{P}(\text{O})(\text{OEt})_3\}_2]$ (**4**) and $[\text{Ru}(\kappa^1\text{-OTf})(\eta^5\text{-C}_5\text{Me}_5)\{\text{P}(\text{O})\text{Ph}_3\}-\{\text{P}(\text{O})(\text{OMe})_3\}]$ (**5**).

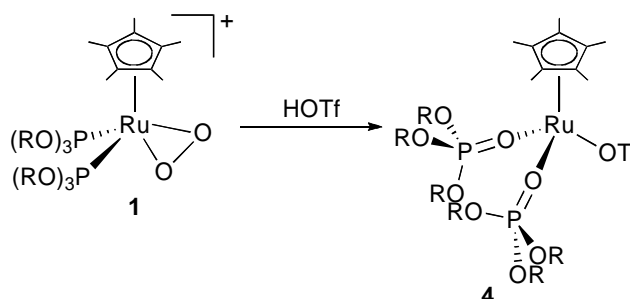


Figure 2: $R = \text{Me}, \text{Et}$

A reaction path for the formation of complexes **4** and **5** is proposed by DFT studies. Besides the phosphate complex **4**, protonation of dioxygen complex **1** under a $\text{CH}_2=\text{CH}_2$ atmosphere (1 atm) afforded acetic acid.

All the complexes were characterised spectroscopically (IR and NMR) and by X-ray crystal structure determination of complexes **1**, **2** and **3**.